

ORIGINAL RESEARCH PAPER

Energy matrices and life cycle conversion analysis of N-identical hybrid double slope solar distiller unit using Al₂O₃ nanoparticle

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ABSTRACT

In the current study, 25% incorporating PVT hybrid CPC collector double slope solar still is using Al₂O₃ nanoparticles underwent energy matrices analysis and life cycle conversion efficiency (LCCE). With the aid of an analytical program fed into MATLAB, the analysis is conducted on an annual basis based on the atmospheric conditions in New Delhi. The IMD in Pune, India, provided weather input data needed for the numerical computations. The average annual energy output will be calculated using energy and exergy and later on evaluated EPT, EPF, and LCCE. This will reveal that the average annual yield is 8.5%, the average energy payback time is 16.16%, the average energy payback factor is 13.91%, and the average life cycle cost conversion efficiency is 7.15% higher compared to the previous research. Therefore, it is obvious the proposed system is better based on the following parameters i.e. (i). annual yield, (ii). energy matrices such as EPT, EPF, and efficiency of life cycle cost (LCCE). The proposed hybrid is a self-sustainable system that can also meet the future requirements of potable water as well as electricity.

Keywords: Energy matrices, Energy payback time, Energy production factor, Life cycle conversion efficiency, Al₂O₃ nanoparticles

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INTRODUCTION

We can access a very low percentage of water from the ground. Therefore, there is a need to develop potable water and self-sustainable systems. Water purification is required due to polluted water to freshwater throughout the world. Consumption of polluted water is increasing death rates by increasing diseases in human beings. In present days, system availability is not self-sustainable. Electricity is needed which generates it causes pollution. Therefore, the better solution is a

renewable energy source that can reduce the potable water problem. Lawrence and Tiwari [1] developed the empirical relations for the inside coefficients of heat transfer from the natural flow with a heat exchanger in a solar distiller unit. Popiel and Wojtkowiak [2] studied the thermo-physical properties of the base fluid. Pak and Cho [3] evaluated various correlations for different properties. G. N. Tiwari [4] studied the fundamental design of the solar still. Hwang et al. [5] analyzed the heat transfer coefficient for Al₂O₃ nanofluids.

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Barden [6] improved the thermo-physical properties of the base fluid; the heat transfer coefficients could also be improved. Due to their superior thermo-physical characteristics, nanoparticles are easily deferred. The nanofluids are developing fluids with extremely quick heat transfer properties. Additionally, the base fluid's qualities could be enhanced by customizing the size and shape. Tiwari and Tiwari [7] expressed few merits of solar distillers over other distillation technologies such as filters, membranes, and batteries, no definitive resource of energy, and primarily low investment. Ho et al. [8] numerically analyzed nanofluids for natural convection in a square enclosure: effects due to uncertainties of viscosity and thermal conductivity. Otanicar and Golden [9] analyzed the enviro economic aspect of solar collectors using nanofluid and found it neutralizes 74 kg for a life span of 15 years. Patel et al. [10] found the thermal conductivity of nanofluids. Singh et al. [11] theoretically investigated entropy generation for nanofluids. Elzen et al. [12] analyzed emission reductions, abatement costs, and carbon prices. Khanafer and Vafai [13] This work presented the thermophysical properties of nanofluids. Khullar and Tyagi [14] analyzed and reported emissions of 103 kg approx./ household/year reduced for a solar heating device for nanofluids. Faizel et al. [15] analyzed based on the cost of flat plate collector (FPC) using tin oxide, copper oxide, titanium oxide, and aluminum oxide) nanofluids. It is discovered that the high density, low specific heat, and thermal conductivity of CuO nanofluid are more appropriately attributed to its performance. Liu et al. [16] have evaluated the economic analysis of the integrated solar distiller unit of the evacuated tube. Kabeel et al. [17] analyzed the sole inclined solar distiller unit with vacuum as a water-based nanofluid. Elango et al. [18] analyzed practically single-slope solar distillers as thermal energy, exergy, and productivity using different nanofluids. Omara et al. [19] analyzed the performance of corrugated wick type and simple solar distiller units using nanofluids. Tiwari et al. [20] analyzed experimentally active solar distillers that exergoeconomic and environmental economic using water-based nanofluid the photovoltaic thermal flat plate collector is met potable water requirements daily. Environmental damage has been estimated to cost \$6.29 annually. Sharon and Reddy [21] analyzed the annual economic performance of an active solar distiller loaded with

saline water. Sahota et al. [22, 24] analyzed the passive double slope solar distiller unit performance using nanofluids and concluded that the aluminum oxide-based nanofluid gives better performance than others. Singh et al. [23] analyzed the energy matrix and existence cycle conversion efficiency for conventional single and double slope distiller units and found 0.144. and 0.137 per unit cost, respectively, and exergoeconomic parameters. Singh and Tiwari [25] analyzed the energy matrices and life cycle cost of an active partly PVT-CPC solar distiller. Shashir et al. [26] analyzed the performance of nanoparticles like copper oxide and graphite micro-flakes on solar distiller units with different cooling on the cover of toughened glass. It is concluded the solar yield increases and copper oxide 47.8% and 57.6%. Sahota et al. [27] studied the performance of PVT-FPC double slope solar distiller unit with or without helical coil heat exchanger using nanofluid and found water-based nanofluid performance was better with a heat exchanger. Saleha et al. [28] analyzed the effect of solvent and found it effective in solar distiller units. Chen et al. [29] analyzed that experimentally found the stability of weak luminous was very good with nanofluid in solar distiller unit and the effect of brackish water's constancy, ocular and thermal properties using nanofluid feasible. Mahian et al. [30] studied and found a significant temperature lower than 50 °C, in a heat exchanger and found a 2 times greater amount of water than without a heat exchanger. Additionally, water nanoparticles improve evaporation at low temperatures. It is crucial to assess the cost-effectiveness of renewable energy systems based on payback. Sahota et al. [31] analyzed environmental economics and exergoeconomics for passive double slope solar distiller with water-loaded nanofluid (CuO, Al₂O₃, TiO₂) and found payback time of energy of the system is low and the cost of environmental per annum is higher on mitigation with nanofluid. Singh and Tiwari [32] analyzed the augmentation in energy matrices of N-PVT-FPC partly double slope solar distiller. Joshi and Tiwari [33] analyzed single slope Nth-identical photovoltaic thermal compound parabolic concentrator collector N-PVT-CPC. Dharamveer et al. [34] reviewed nanofluid-loaded desalination. Kumar and Singh [35] analyzed the Energy and exergy of active solar stills using a compound parabolic concentrator. Shanker, et al. [36] analyzed the performance of the C.I. engine using biodiesel fuel by modifying

injection timing and injection pressure Anup et al. [37] analyzed using FEA of refrigerator compartment for optimizing thermal efficiency. Kumar and Singh. [38] optimized thermal behavior of a small heat exchanger. Zhang et al. [39] presented in the area of sustainable energies focuses on utilizing green and clean technologies. Dhivagar et al. [40] analyzed single slope grate crude shrewd solar distiller units for energy, exergy, and economic aspects. Dharamveer and Samsheer [41] studied the active and passive solar still behavior on energy matrices and enviroeconomics. Arora et al. [42] analyzed double slope solar distiller N-PVT-CPC using carbon nanotubes for water generation. Dharamveer et al. [43] analytically studied Nth identical photovoltaic thermal (PVT) compound parabolic concentrator (CPC) active double slope solar distiller with a helically coiled heat exchanger using C_{60} Nanoparticles. Dharamveer, et al. [44] analyzed an N-identical active single-slope solar distiller with a helically coiled heat exchanger using CuO nanoparticles. Kumar and Singh, [45] compared single-phase microchannels for heat flow Experimental and using CFD. Subrit and Singh. [46] analyzed thermal of coal and waste cotton oil liquid produced by pyrolysis of diesel engine fuel was carried out by. Shahsavari et al. [47] Compared energy, exergy, environmental, exergoeconomic, and enviro economic analysis of building integrated photovoltaic/thermal, earth-air heat exchanger, and hybrid systems. Numerous studies on passive and active solar stills have been conducted, according to the current literature survey. However, not much investigation on active solar still filled with water-based nanofluids was studied. Based on energy and exergy, Dharamveer et al. studied a hybrid double slope. No scholars have examined the economic, and environmental, using nanofluid. Furthermore, no studies have been conducted for CPC, ETC double slope basin type solar distiller using nanofluid. Thus, the proposed study will examine the impacts of active solar still double slope with CPC and filled with water containing Al_2O_3 nanofluid on energy matrices based such as energy payback time, energy payback factor, life cycle cost conversion efficiency, and productivity, of solar desalination systems will be thoroughly examined. The effectiveness of the suggested approach will also be evaluated in comparison to the findings of past studies.

MATERIALS AND METHODS

To determine the following objectives the methodology is adopted as energy matrices analysis of hybrid solar distiller basin type double slope with heat exchanger using Al_2O_3 nanoparticles.

System Description

Working of double slope solar desalination incorporating PVT with CPC collector using nanoparticles (N-PVT-CPC-DS-HE) is shown. Representation of solar still is followed the greenhouse principle. The parameters used for the distiller unit are given in Table 1 and Table 2. The basin of the solar distiller is connected in series with N-CPC and incorporated with a helically coiled heat exchanger. Solar radiation received on the glass cover is transferred to the water surface and thereafter absorbed by the surface. Later on, reflected water received heat on the top cover and rest portion and then move to the liner where the maximum amount absorbed and liner temperature increased and transfer this heat to water. Thus the temperature of water in the basin is increased and water gets evaporated. Collectors also heat the water in the basin. Water is heated and evaporates in this manner. The distillate trickles forward to the passage attached to the bottom side as the vapor reaches the interior face of the condenser, where film-wise condensation takes place. Then, the beaker receives the distilled water that was siphoned off.

Fig. 1 represents 25% PVT incorporating hybrid solar still. Collectors are put south facing at an angle of 45° which are connected in series as the input of the second collector is attached to the first collector output. Radiation that falls on the collector directly gets absorbed and beam irradiation is reflected on the parabolic concentrator. Similarly, irradiation falls on PV modules that generate electricity which is further used for operating pumps (D.C) and access energy can be further used for any electrical appliances according to need. Table 2 represents the specifications of the proposed system. The inclination angle of the system is 30° and the orientation of the system is the southern face. The basin liner is absorbed maximum radiation which falls on the glass cover. The basin fluid gets heat from the heat exchanger through Al_2O_3 nanoparticles. As per a prior study, it is obvious that a helically coiled heat exchanger is more effective than any other design. The Al_2O_3 nanoparticles exchange more heat in active solar still because it covers more

Table 1. Specifications of double slope PVT-CPC-DS active solar distillation system [43]

Hybrid solar distiller unit			
Components		Particulars	
Basin length		2.0 m	
Basin width		1.0 m	
Tilting glass cover		15°	
Basin (smaller side) height		0.2 m	
Body's material		G.R.P	
Stand's material		G.I.	
Covers		Glass	
Collector facing		south	
Tilting glass thickness		0.004 m	
K_g		0.816 W/mK	
Insulation width		0.1 m	
Thermal conductivity of insulation		0.166 W/mK	
Hybrid collector			
Component's name	Specifications	Component's name	Specifications
Kinds of collector	Tube kinds	Area of aperture	2 m ²
Receiver area	1.0 m ²	Aperture's module area	1.0 m ²
Collector's plate thick	0.0020 m	Aperture's receiver area	1.5 m ²
Copper's tube thick	0.00056 m	Module's receiver area	0.25 m ²
Copper's tube length	1.0 m	Collector's receiver area	0.75 m ²
Packing coefficient	0.8	No of collectors	N=4
D.C. motor	12V, 40 Watt	Pipe dia.	0.0125 m
Inclination of CPC	30°	Inclination of FPC	30°
Under glass effective area	0.75 m ²	Under PV module effective area	0.660 m ²
No. of the cell (solar)	36	Single-cell (Solar)	0.007 m ²
Basin area	2 m ²	m_f	0.02 kg/s
Heat exchanger (copper coiled)			
No. of turns	12	Tube coil dia	0.0125 m
Length of heat exchanger	1.937 m	Heat exchanger coil dia	0.045 m



Table 2. Parameters are used for calculations [43]

Specifications	Numerical values	Specifications	Numerical values
α_g	0.050	A_{am}	0.5
α_b	0.5861	A_{ac}	1.5
α_{bf}	0.82	L_p	0.0020
α_c	0.90	K_m	64 W/mK
α_p	0.80	K_p	64 W/mK
β_c	0.89	L_i	0.1m
K_g	0.8160 W/mK	h_i	5.7 W/m ² K
K_b	0.035 W/mK	h_o	9.5 W/m ² K
K_p	0.166 W/mK	U_{tcp}	5.5451 W/m ² K
L_g	0.004	U_{tca}	15.03 W/m ² K
L_c	0.005	U_{tpa}	5.56 W/m ² K
L_i	0.1	U_{Lm}	9.03 W/m ² K
β_o	0.0045/K	U_{LC}	5.43 W/m ² K
x	0.33m	PF_1	0.2695
σ	$5.67 \times 10^{-8} (W/m^2K^4)$	PF_2	0.9398
τ_g	0.95	PF_c	0.977
$F/$	0.968	ϵ_g	0.95
η_o	0.15	ϵ_{bf}	0.95

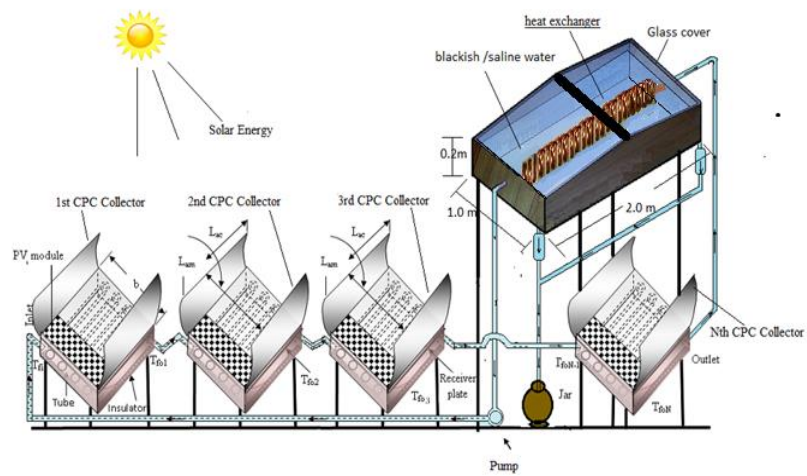


Fig. 1 active double slope solar still 25% incorporating N-PVT-CPC-DS-HE

surface area due to increasing volume due to heat exchanger. The system gets thermal energy via a combination of double-slope basins incorporating with PVT-CPC collector unit and it absorbs heat externally from a collector and internally from a basin. Basin water temperature increases via heat exchanger nanoparticles. Ultimately by releasing latent heat the vapor gets condensed, and collected at the lower end of the inclined glass cover of the basin.

This sort of system produced electricity as well as potable water. The other quality of this sort of system is i.e. low maintenance, easy to install, and useful for large and small demands of potable water as well as industrial purpose. Lot of advantages of such a system but in this work, we emphasize producing potable water only therefore according to making the system self-sustainable the PVT is provided otherwise 50%, 75%, and 100% can also be used.

Though, the proposed system-A is compared with the previous System-B based on basin water temperature, inside glass cover temperature, water outlet temperature, the overall thermal energy of collectors, overall exergy, overall electrical exergy, and potable water amount. Sedimentation possibility in nanofluid is more. Therefore nanoparticles size leads to a change in aggregation. The value of size matters to change in aggregation. Later on, more sophisticated equipment is needed to remove it.

Governing equations

To develop the characteristic equation, the following assumptions are:

- i. Constant water level
- ii. Neglected ohmic losses
- iii. No leakage
- iv. Over the entire surface film condensation
- v. Steady-state partially covered active solar stills

The governing equations of the system are as follows:

a. East face

$$\alpha_g I_{SE} A_{gE} + h_{1wE} (T_w - T_{giE}) \frac{A_b}{2} - h_{EW} (T_{giE} - T_{giW}) A_{gE} = U_{cgaE} (T_{giE} - T_a) A_1 \tag{1}$$

b. West face

$$\alpha_g I_{SW} A_{gW} + h_{1wW} (T_w - T_{giW}) \frac{A_b}{2} + h_{EW} (T_{giE} - T_{giW}) A_{gW} = U_{cgaW} (T_{giW} - T_a) A_{gW} \tag{2}$$

On solving Equations (1) and (2)

$$T_{giE} = \frac{A_1 + A_2 T_w}{P} \tag{3}$$

$$T_{giW} = \frac{B_1 + B_2 T_w}{P} \tag{4}$$

$$T_{goE} = \frac{\frac{k_g T_{giE} + h_{1gE} T_a}{L_g} + h_{1gE} T_a}{\frac{k_g}{L_g} + h_{1gE}} \tag{5}$$

$$T_{goW} = \frac{\frac{k_g T_{giW} + h_{1gW} T_a}{L_g} + h_{1gW} T_a}{\frac{k_g}{L_g} + h_{1gW}} \tag{6}$$

The unknown terms, A₁, A₂, B₁, B₂, and P in Equations 3, 4, 5, and 6 are mentioned in Appendix A,

c. Basin liner

$$\alpha_b (I_{SE} + I_{SW}) + 2h_{bw} (T_b - T_w) + 2h_{ba} (T_b - T_a) \tag{7}$$

d. Water equation

$$m_f C_f \frac{dT_w}{dt} = \alpha_w (I_{SE} + I_{SW}) \frac{A_b}{2} + 2h_{bw} (T_b - T_w) \frac{A_b}{2} - h_{1wE} (T_w - T_{giE}) \frac{A_b}{2} - h_{1wW} (T_w - T_{giW}) \frac{A_b}{2} + Q_{un} \tag{8}$$

e. Energy balance equations for heat exchange and

base fluid can be written for solar distillers as:

$$m_f C_f \frac{dT_w}{dx} d_x = - (2\pi r_{11} U) (T_{HE} - T_w) d_x \tag{9}$$

Applying the limits T_w at (x = 0) =

T_{w0N} and T_w at (x = L) = T_{w1} on solving,

$$T_{wi} = T_{HE} \left[1 - \exp\left(\frac{-2\pi r_{11} UL}{m_f C_f}\right) \right] + T_{w0N} \exp\left(\frac{-2\pi r_{11} UL}{m_f C_f}\right) \tag{10}$$

Where $U = \left[\frac{1}{h_{bf}} + \left(\frac{r_{11}}{k_1} \right) \ln\left(\frac{r_{22}}{k_1} \left(\frac{1}{h_{bf}} \right) \right) \right]^{-1}$

f. Energy balance for water collector N-PVT-CPC

$$T_{w0N} = \left[\frac{(AF_R(\alpha\tau))_1 (1 - K_p^N)}{m_f C_f (1 - K_p)} \right] I_b + \left[\frac{(AF_R(UL))_1 (1 - K_p^N)}{m_f C_f (1 - K_p)} \right] T_a + T_{wi} K_m^N \tag{11}$$

$$T_{w0N} = \left[\left[\frac{(AF_R(\alpha\tau))_1 (1 - K_p^N)}{m_f C_f (1 - K_p)} \right] \left(\frac{1}{(1 - e^{-2K_m^N})} \right) \right] I_b +$$



$$\left[\frac{(AF_R(UL)1)(1-K_p^N)}{m_f C_f (1-K_p)} \left(\frac{1}{(1-e^z K_m^N)} \right) \right] T_a + T_{HE} \left(\frac{(1-e^z) K_m^N}{(1-e^z K_m^N)} \right) \quad (12)$$

Using this relation heat gain computed

$$Q_{uN} = m_f C_f (T_{woN} - T_{wi}) \quad (13)$$

The outlet temperature of water computed by these two equations 3.12 and 3.13,

$$T_{HE} > T_{wi} > T_w$$

$$Q_{uN} = \left[\left(\frac{(AF_R(\alpha)1)(1-K_p^N)}{(1-K_p)} \right) \left(\frac{1}{(1-e^z K_m^N)} \right) \right] I_b + \left(\left(\frac{(AF_R(UL)1)(1-K_p^N)}{(1-K_p)} \right) \left(\frac{1}{(1-e^z K_m^N)} \right) \right) T_a + m_f C_f \left(T_{HE} \left(\frac{(1-e^z) K_m^N}{(1-e^z K_m^N)} \right) - T_{wi} \right) \quad (14)$$

By putting

T_{giE} , T_{giW} , $2h_{bw} (T_b - T_w)$, and Q_{uN} from Equations (3),(4), and (9) in Equation (15)

$$\frac{dT_w}{dt} = -a_2 T_w + f_2 (t) \quad (15)$$

The energy and exergy analysis has been done, respectively, based on the thermodynamics laws for the given entities.

$$E_{hourlyEn} = [h_{1wE} (T_w - T_{giE}) + h_{1wW} (T_w - T_{giW})] (A_b) \quad (16)$$

$$E_{hourlyEx} = \{h_{1wE} [(T_w - T_{giE}) - (T_a + 273)] + h_{1wW} [(T_w - T_{giW}) - (T_a + 273)]\} A_b \quad (17)$$

The constant 0.933 represents the solar irradiation exchange constant, per hour water production for the proposed system have been calculated using the following equation.

$$M_w = \frac{q_{ew}}{L_v} 3600 = \frac{h_{ew} (T_w - T_g)}{L_v} 3600 \quad (18)$$

Latent heat of vaporization is expressed as [43]:

$$L_v = 3.162510^6 + [1 - (7.61610^{-4} T_v)] \text{ For } T_v > 70 \text{ }^\circ\text{C}$$

$$L_v = 2.493510^6 [1 - (9.477910^{-4} T_v)] + 1.313210^{-7} (T_v^2) - 4.797410^{-3} (T_v^3) \text{ For } T_v < 70 \text{ }^\circ\text{C}$$

For $T_v < 70 \text{ }^\circ\text{C}$

Thermophysical properties of vapor, base fluid from references [31-34], and Nanoparticles (Table - 3) [37- 42].

Analysis based on matrices and cost conversion based on the life span of hybrid active double slope solar distiller unit using Al_2O_3 nanoparticles -

Energy matrices inform about the time of energy payback (EPT), energy payback factor (EPF), and life cycle conversion efficiency for the lifetime period (LCCE) [27].

Energy Payback Time (EPT) [27]

The time duration needed to recover total exhausted energy in manufacturing material can be expressed as

$$EPT_{(e)} = \frac{\text{Embodied Energy } (E_{in})}{\text{Annual energy output } (E_{out})} \quad (19)$$

$$EPT_{(ex)} = \frac{\text{Embodied Energy } (E_{in})}{\text{Annual exergy output } (E_{out})} \quad (20)$$

Energy Production Factor (EPF) [27]

It is reciprocal of EPT an ideal value of EPF on an annual basis to express the overall performance of solar still can be expressed as

$$EPF_{(e)} = \frac{\text{Overall Energy Output } (E_{out})}{\text{Embodied Energy } (E_{in})} \quad (21)$$

$$EPF_{(ex)} = \frac{\text{Overall Exergy Output } (E_{out})}{\text{Embodied Energy } (E_{in})} \quad (22)$$

Lifecycle Conversion Efficiency (LCCE) [27]

$$LCCE_{(e)} = \frac{E_{out} \times n - E_{in}}{E_{sol} \times n} \quad (23)$$

Where, E_{sol} represents annual solar energy (kWh) and n is the overall lifetime period.



Table 3. Thermophysical properties of Al₂O₃ nanoparticles [43]

Nanoparticles (NPs)	Density k _g /m ³	Thermal conductivity k _p (W/mK)	Specific heat C _p (j/k _g K)
Al ₂ O ₃	6.3*10 ³	17.6	550

Table 4. Daily, Monthly, and Annually yield of the proposed system

Month	whether condition type (a)		whether condition type (b)		whether condition type (c)		whether condition type (d)		Monthly yield				
Jan	14.70	3	44.09	12.78	8	102.21	4.17	11	45.86	1.21	9	10.85	203.01
Feb	13.82	3	41.47	13.44	4	53.75	4.57	12	54.81	1.42	9	12.78	162.82
Mar	18.20	5	90.98	18.10	6	108.63	9.02	12	108.19	5.00	8	40.00	347.81
Apr	20.51	4	82.03	21.53	7	150.68	11.72	14	164.12	10.90	5	54.49	451.32
May	21.98	4	87.92	19.45	9	175.01	15.07	12	180.81	11.00	6	66.00	509.74
Jun	12.49	3	37.48	18.97	4	75.89	12.25	14	171.56	7.83	6	47.01	331.94
Jul	15.19	2	30.38	15.19	3	45.57	11.51	10	115.12	6.70	17	113.97	305.04
Aug	15.27	2	30.53	15.47	3	46.40	9.65	7	67.52	5.78	19	109.73	254.19
Sep	16.46	7	115.23	16.03	3	48.10	12.18	10	121.79	7.66	10	76.64	361.76
Oct	15.11	5	75.53	10.88	10	108.84	8.42	13	109.44	4.36	3	13.07	306.89
Nov	14.81	6	88.88	10.22	10	102.16	4.12	12	49.50	3.71	2	7.41	247.95
Dec	14.79	3	44.36	9.14	7	64.01	4.92	13	64.01	1.44	8	11.53	183.92
Annual yield in (kg)													3666.39

Economic analysis is required to determine of hybrid active double slope solar still using Al₂O₃ nanoparticles-

It is economically feasible for System-A and System-B for both systems.

Capital cost

Tables 5 and 6 provide the system's fabrication cost.

System's Lifespan

It is considered for 15, 20, and 30 years.

Salvage value (S)

$$\text{Salvage value (s)} = 0.2 \times \text{Principal capital (PCC)} \quad (24)$$

PCC stands for principal capital cost

Cost of annual salvaging (ASC)

$$\text{Annually salvage} = S \times \text{shrinkage in a fund (SFF)} \quad (25)$$

SFF is a factor used for shrinkage

Yearly maintenance (AMC)

$$\text{Yearly maintenance (AMC)} = 0.15 \times \text{FAC} \quad (26)$$

FAC stands for the first annual cost.

Factor for capital recovery (CRF).

At a fixed rate of interest, it shows the present cost as a constant annual cost across time.

$$\text{CRF} = \frac{i(1+i)^n}{(1+i)^n - 1} \quad (27)$$

Shrinking fund factor

$$\text{SFF} = \frac{i}{(1+i)^n - 1} \quad (28)$$

Firstly annual cot estimated

$$\text{FAC} = \text{PCC} \times \text{CRF} \quad (29)$$

Net annual cost estimated

$$\text{TAC} = \text{FAC} + \text{AMC} - \text{ASC} \quad (30)$$

Cost of distillate per kg obtained

$$\text{Cost/kg} = \frac{\text{TAC}}{\text{yield in life}} \quad (31)$$



Table 5. Embodied Energy of different components of system-A (N-PVT-DS-CPC-HE) and system-B (N-PVT-DS-FPC-HE)

Name of component	Embodied energy (kWh)	
	N-PVT-FPC-DS-HE (Previous system)[30]	N-PVT-CPC-DS-HE (Present system)
FRP body of DS	755.61	755.61
GI angle Iron	416.4	416.4
Glass cover	180.5	180.5
FPC (N=4)	2209.92	-
CPC (N=4)	-	3279.41
PV (glass-glass)	980	980
Copper heat exchanger	25.83	25.83
Nanoparticles (CuO)	17.82	17.82
Others	20	20
Total EE of system	4606.08	5675.57

Table 6. Capital investment and cost of different components of system-A (N-PVT-DS-CPC-HE) and system-B (N-PVT-DS-FPC-HE)

Parameters	N-PVT-FPC-DS-HE [32]		N-PVT-CPC-DS-HE	
	Cost	Previous system	Cost	Present system
	₹	\$	₹	\$
FRP body	10200	139.135	10200	139.135
Glass cover 2.05	1600	21.825	1600	21.825
Iron stand	1000	13.641	1000	13.641
Inlet /outlet nozzle	200	2.728	200	2.728
Iron clamp	250	3.410	250	3.410
Gaskets	200	2.728	200	2.728
Silicon gel	200	2.728	200	2.728
PVT-FPC (N=4) 8500	34000	463.784		0.000
PVT-CPC (N=4) 9000		0.000	36000	491.06
Motor and pump	1200	16.369	1200	16.369
Helically coiled heat exchanger	466.25	6.360	466.25	6.360
Fabrication and other cost	6000	81.844	6000	81.844
100 gmsAl ₂ O ₃ nanoparticles	7425	101.282	7425	101.282
Total cost of	62741.25	855.83	64741.25	883.11

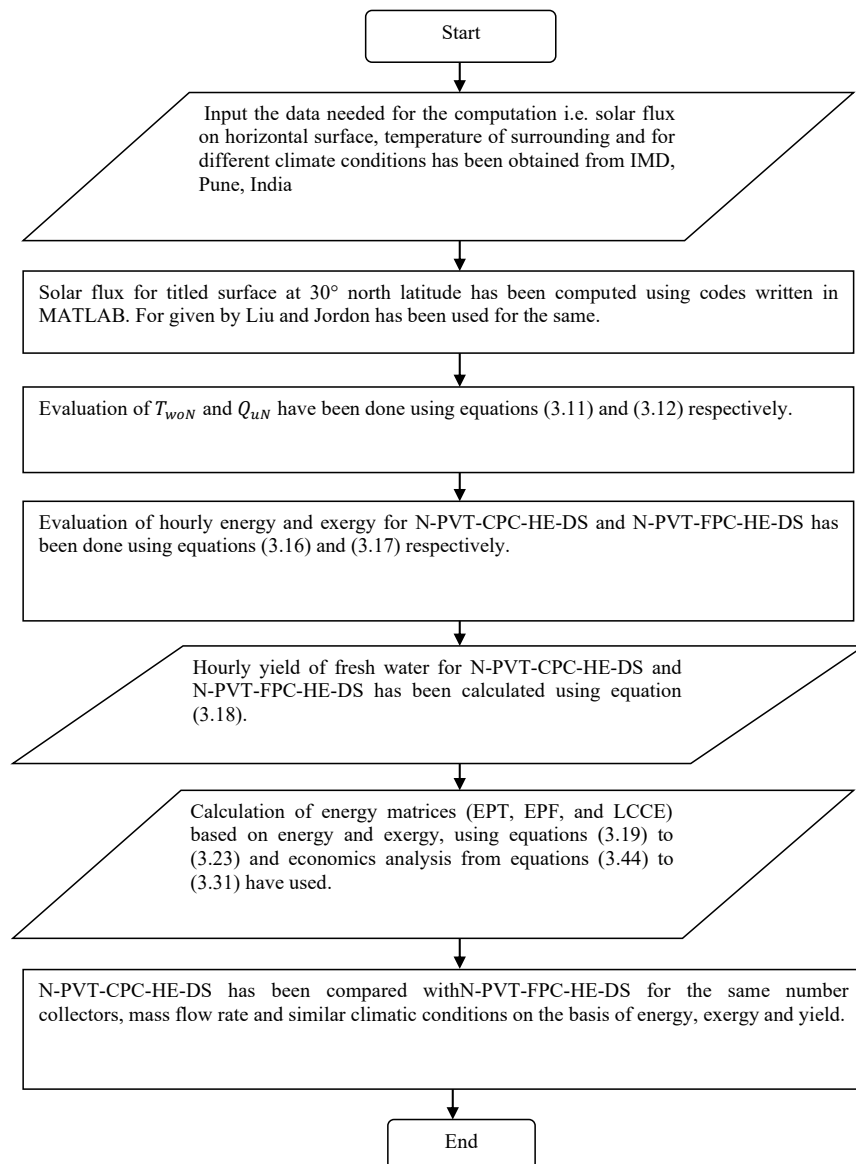


Fig. 2 Flow chart of the methodology adopted

Methodology to be Adopted

The following steps are included in the approach used to study the suggested system (Fig. 2):

Step-I

The proposed systems for the yearly are calculated using the Lui-Jordon formula for global and beam irradiation. Calculate daily solar radiation further by multiplying the number of days given in a month by the number of clear, hazy, hazy, cloudy, and cloudy days.

Step-II

The temperature of basin water is calculated based on hourly, monthly, and annual data, and

all settings are tuned to maximize the collector's output temperature.

Step-III

Energy matrices such as efficiency of life cycle cost, Factor energy payback, energy payback time, and productivity have been evaluated.

Step-IV

Comparing proposed systems to the prior system using numerically computed values.

RESULT AND DISCUSSION

The solar irradiation on a flat area and surroundings temperature have been computed using

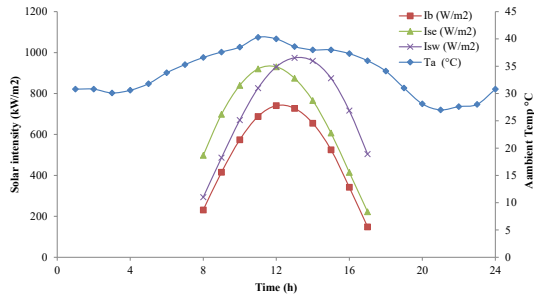


Fig. 3 Hourly variation in solar irradiation and ambient temperature for typical day of May

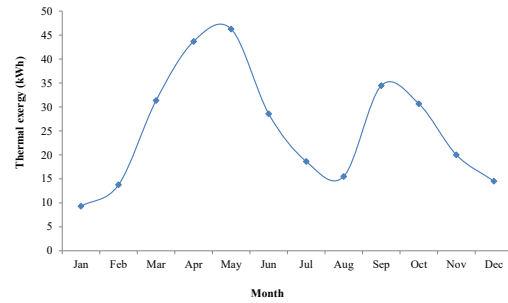


Fig. 4 Shows the thermal exergy of the proposed system-A

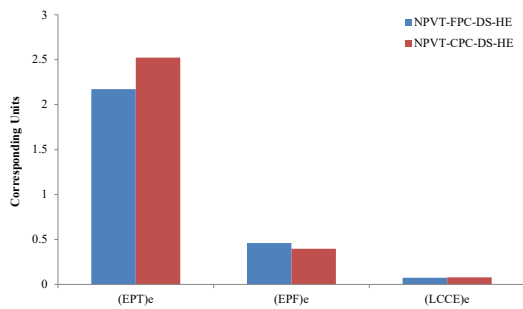


Fig. 5 Shows EPT, EPF, and LCCE for proposed and previous system for 15 years

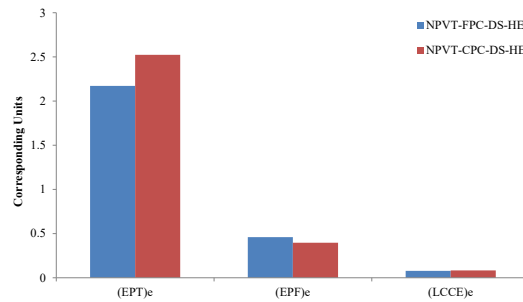


Fig. 6 Shows EPT, EPF, and LCCE for proposed and previous system for 20 years

IMD Pune data, India. By entering the pertinent data in MATLAB, the Liu and Jordan formula may be used to determine how much radiation was applied to N-PVT-CPCs. The values of beam radiation I_b , solar radiation Eastside ISE, Westside ISW, and ambient temperature T_a in kW/m^2 and $^\circ\text{C}$, respectively, are shown in Fig. 3. Below Fig.4 shows a variation of the thermal exergy month-wise of proposed system-A.

Energy matrices and a life cycle cost conversion analysis are required for the hybrid active double slope solar distiller unit

Embodied energy (E_{in}), the conversion efficiency of the life cycle (LCCE), energy payback factor (EPF), and energy payback time for (N-PVT-DS-FPC-HE) and (N-PVT-DS-CPC-HE) are shown in Fig. 5, Fig. 6, and Fig. 7 for 15, 20 and 30 years respectively.

The number of energy matrices based on energy and exergy for proposed system-A is found that system-A is better to system-B. EPT based on energy and exergy is 16.16% and 17.84% higher, respectively. EPF based on energy and exergy is relatively less 13.91% and 14.88%, respectively. LCCE based on energy and exergy is appreciably greater for system-A than system-B: 5.55%, 6.38%, and 7.15%

for 15, 20, and 30 years of life considered, respectively.

Economic analysis is required to determine whether hybrid active solar still uses Nanoparticles (Al_2O_3)

It is economically feasible for system-A and System-B annually. Total yearly cost, fixed annual cost, yearly maintenance cost, and yearly water generation for system-A and system-B are shown during 30 years, at respective interest rates of 1%, 3%, and 5% respectively are represented in Fig. 8, and Fig. 9.

The cost of distillation relies on the rate of interest, as results over 30 years are shown in Fig.8. While the cost of distillate will reduce as system life increases, the price of distillate will climb annually as interest rates rise. For system-A and system-B, the cost of distillate is 0.69, 0.96, and 1.27 (/kg), and 0.72, 1.012, and 1.33 (/kg), respectively, over 30 years at interest rates of 1%, 3%, and 5%. According to the cost of distillation, it is concluded that the proposed system-A performs better than system B.

CONCLUSION AND FUTURE SCOPE

The proposed system and previous system performance have been studied using characteristic



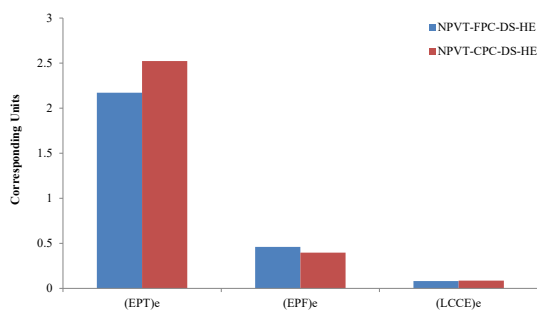


Fig. 7 Shows EPT, EPF, and LCCE for proposed and previous system for 30 years

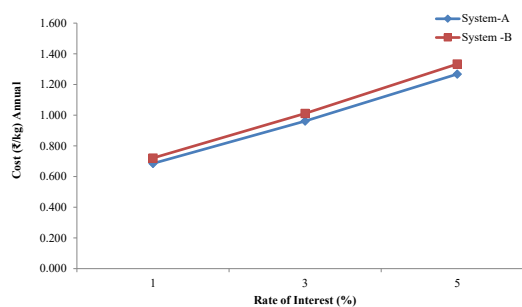
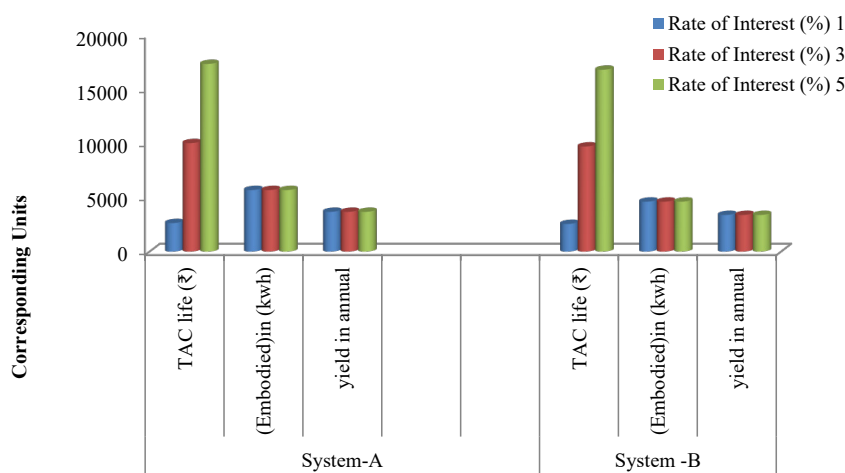


Fig. 8 Represents cost of distillate vs Interest rate



Systems

Fig. 9 Shows comparison for system-A and system-B based on TAC, Embodied Energy and Yield for 30 years

equations and Al₂O₃ nanoparticles and found better than the previous system.

Conclusions

The following conclusions are made by the annual analysis of the proposed systems with Al₂O₃ nanoparticles.

1. The proposed system-A gives better annual performance than system-B based on thermal exergy, energy, yield is 8.5%, and productivity 5.17% greater with Al₂O₃ nanoparticles.
2. Al₂O₃ nanoparticles-based system-A gives better results on annual performance and economics as compared to system-B.
3. Based on thermal energy, thermal exergy, energy, and exergy-based, Energy payback time (EPT) is 16.16%, Energy payback factor (EPF) is 13.91%, Efficiency of life cycle cost (LCCE) is 7.15% greater than the previous research, it is found that system-A outperforms system-B (previous).
4. System-A has a lower distillation cost than System

B (previous). The yearly cost of distillate for systems A and B is 0.69, 0.96, and 1.27 (/kg), and 0.72, 1.012, and 1.33 (/kg), respectively, based on a 30-year basis at interest rates of 1%, 3%, and 5%. According to distillate cost, it is discovered that system-A performs better than system B.

Future scope

This work can be further expanded using research with PCM material in CPC up to a specific level.

1. Energy and exergy can be studied for different nanoparticles
2. energy matrices, EPT, EPF, and LCCE can be studied for different nanoparticles
3. Research on the environmental and economic benefits of various nanoparticles is possible. Different sizes, and shapes of nanoparticles can be investigated.
4. The partial covered 50%, 75%, and 100% can be studied.



Nomenclature

A_{ba} basin surface, in (m^2)	h_{1wE} eastside water to glass cover coefficient of heat transfer, in ($W/m^2 K$)
A_{ca} flat plate collector area under glazing, in (m^2)	h_{1wW} westside water to glass cover coefficient of heat transfer, in ($W/m^2 K$)
A_{gE} eastside glass cover area, in (m^2)	I_b on the collector solar irradiation, in (W/m^2)
A_{gW} westside glass cover area, in (m^2)	I_{SE} eastside over-glass cover solar irradiation, in (W/m^2)
A_m PVT area, in (m^2)	I_{SW} westside over glass cover solar irradiation, in (W/m^2)
C_p specific heat NPs, in (J/kgK)	K_g in glass heat conductivity($W/m K$)
C_{bf} base fluid specific heat, in (J/kgK)	K_p in absorbing plate heat conductivity ($W/m K$)
C_{nf} specific heat nanofluid, in (J/kgK)	k_p in nanoparticles, heat conductivity($W/m K$)
D_i FPC tube dia., in (m)	K_{nf} in nanofluid heat conductivity ($W/m K$)
d_p Nps dia., in (nm)	K_{bf} in base fluid heat conductivity($W/m K$)
F' the factor of collector efficiency	L length of heat exchanger coil, in (m)
h_i coefficient of heat transfer glazing to absorbing plate, in ($W/m^2 K$)	L_c glazing length collector, in (m)
h_o heat transfer coefficient top of PVT to ambient air, in ($W/m^2 K$)	L_i the thickness of insulation, in (m)
h_{pw} heat transfer coefficient blackened plate to fluid, in ($W/m^2 K$)	L_g cover of glass thickness, in (m)
h_{bw} heat transfer coefficient liner to ambient, in ($W/m^2 K$)	L_m length of PVT, in (m)
h_{ba} coefficient of heat transfer from basin to ambient, in ($W/m^2 K$)	L_p absorbing plate thickness, in (m)
h_{CPC} coefficient of heat transfer convective in CPC, in ($W/m^2 K$)	M_w water mass, in (k_g)
h_{HE} in heat exchanger convective heat transfer coefficient, in ($W/m^2 K$)	M_{ew} yield, in (k_g)
h_{rwgE} eastside, water-to-glass heat transfer coefficient radiative, in ($W/m^2 K$)	m_f water flow rate, in (k_g/s)
h_{rwgW} westside, water-to-glass heat transfer coefficient radiative, in ($W/m^2 K$)	PF_1 penalty factor 1
h_{cwgE} eastside, water-to-glass heat transfer coefficient radiative, convective in ($W/m^2 K$)	PF_2 penalty factor 2
h_{cwgW} westside, water-to-glass heat transfer coefficient radiative, convective in ($W/m^2 K$)	PF_3 penalty factor 3
h_{ewgE} eastside, water-to-glass heat transfer coefficient radiative, evaporative in ($W/m^2 K$)	PF_c by glass cover for glazed portion, a penalty factor
h_{ewgW} westside, water-to-glass heat transfer coefficient radiative, evaporative in ($W/m^2 K$)	P_{gi} partially saturated vapor pressure of glass cover, in (N/m^2)
h_{1gE} eastside, coefficient of total heat transfer, in ($W/m^2 K$)	Q_{uN} the heat transfer rate to N identical 25% PVT-CPC connected in series, in (kWh)
h_{1gW} westside, coefficient of total heat transfer, in ($W/m^2 K$)	r_{11} outerdia of helical coiled heat exchanger tube, in (m)
	r_{22} innerdia of helical coiled heat exchanger tube, in (m)
	T_{giE} eastside, inside glass cover temperature, in ($^{\circ}C$)
	T_{giw} westside, inside glass cover temperature, in ($^{\circ}C$)
	T_{goE} eastside temperature of the outside glass cover, in ($^{\circ}C$)

T_{goW} westside temperature of the outside glass cover, in ($^{\circ}C$)
 T_a surrounding temperature in($^{\circ}C$)
 T_c the temperature of solar cell in($^{\circ}C$)
 T_{cN} the average temperature of a solar cell, in ($^{\circ}C$)
 T_{fi} fluid inlet temperature in ($^{\circ}C$)
 T_{bf} in collector base fluid temperature, in ($^{\circ}C$)
 T_{foN} N^{th} collector outlet water temperature in ($^{\circ}C$)
 T_{nf} nanofluid temperature in($^{\circ}C$)
 T_p absorbing plate temperature in ($^{\circ}C$)
 T_s sun temperature in($^{\circ}C$)
 T_v vapor temperature in($^{\circ}C$)
 T_w the temperature of basin water in($^{\circ}C$)
 T_{wo} temperature of water at $t=0$, in($^{\circ}C$)
 ΔT_{DSSS} temperature between nanofluid and base fluid, in ($^{\circ}C$)
 ΔT_{CPC} temperature between nanofluid to base fluid at PVT collectors outlet, in ($^{\circ}C$)
 ΔT_{HE} temperature between nanofluid to base fluid at the heat exchanger, in ($^{\circ}C$)
 ΔT temperature between T_w and $\frac{T_{giE}}{T_{giW}}$ for time(t), in (h)
 U_{ba} sink line to surrounding, coefficient of overall heat transfer, in ($W/m^2 K$)
 U_{ga} condensing cover to surrounding, overall heat transfer coefficient, in ($W/m^2 K$)
 U_{gaE} condensing cover to surrounding, overall heat transfer coefficient eastside, in ($W/m^2 K$)
 U_{gaW} condensing cover to surrounding, overall heat transfer coefficient, in ($W/m^2 K$)
 U_{Lc} glazing to surrounding, overall heat transfer coefficient, in ($W/m^2 K$)
 U_{Lm} module to surrounding, overall heat transfer coefficient, in ($W/m^2 K$)
 U_{tca} solar cell to surrounding, overall heat transfer coefficient, in ($W/m^2 K$)
 U_{tpa} absorbing plate to surrounding, overall heat transfer coefficient, in $W/m^2 K$
 $U_{tc,p}$ solar cell to absorbing plate, overall heat transfer coefficient, in ($W/m^2 K$)
 X characteristic length, solar distiller unit, in (m)

Greek letters

α_g solar energy fraction absorbed by the condensing cover
 α_b solar energy fraction absorbed by basin surface
 α_f solar energy fraction absorbed by the fluid
 α_c solar energy fraction absorbed by solar cell
 β packing factor
 β_p nanoparticles, coefficient of thermal expansion, in (K^{-1})
 β_{nf} nanofluid coefficient of thermal expansion, in (K^{-1})
 β_{bf} base fluid coefficient of thermal expansion, in (K^{-1})
 \emptyset_g nanoparticles fraction of volume, in (%)
 μ_{bf} base fluid viscosity of dynamic, in (Ns/m^2)
 μ_{nf} nanofluid viscosity of dynamic, in (Ns/m^2)
 η_g collectors efficiency, in %
 ρ_p nanoparticles density, in (Kg/m^3)
 ρ_{nf} nanofluid density, in (Kg/m^3)
 ρ_{bf} base fluids density, in (Kg/m^3)
 τ_g transmitted fraction of glass cover

Subscripts

a ambient
 a_n per annum
 b basin
 E eastside
 e_n energy
 e_x exergy
 E_{in} embodied energy input
 E_{out} embodied energy output
 E_{sol} annual solar energy
 f fluid
 g_i condensing cover inside
 g_o condensing cover outside
 i interest rate
 n life period
 p particle
 S_{ol} solar
 th thermal
 v vapor
 W westside

Abbreviation

- AMC maintenance costs annually
- ASC salvage cost annually
- BF base fluid
- CRF capital recovery factor
- C per liter yield price
- CM carbon dioxide mitigation
- DS double slope solar distiller unit
- EPT payback time of energy
- EPF payback factor of energy
- FAC fixed annual cost
- FPC collector, flat plate
- HE heat exchanger
- HTC coefficient of heat transfer
- LCCE efficiency of life cycle conversion
- NF nanofluid
- NP nanoparticle
- PCC primary capital cost
- PVT photovoltaic thermal
- R reflectors
- SFF shrinking fund factor
- S value of future salvage
- TAC total annual cost
- CPC compound parabolic concentrator
- N-PVT-DS-CPC-HE, incorporating PVT-CPC double slope with Nth collector using a heat exchanger (helically coiled)

Appendix A

$$A_1 = C_1 U_1 + C_2$$

$$C_1 = \alpha_g I_{SE} A_{gE} + U_{cgaE} A_{gE} T_a$$

$$C_2 = \alpha_g I_{SW} A_{gE} A_{gW} h_{EW} + U_{cgaW} A_{gE} T_a A_{gW} h_{EW}$$

$$U_1 = h_{1wE} \frac{A_b}{2} + h_{EW} A_{gE} + U_{cgaE} A_{gE}$$

$$A_2 = (h_{1wE} U_2 + h_{1wW} h_{EW} A_{gE}) \frac{A_b}{2}$$

$$U_2 = U_{cgaW} A_{gW} + h_{1wW} \frac{A_b}{2} + h_{EW} A_{gW}$$

$$B_1 = C'_1 U_1 + C'_2$$

$$B_2 = (h_{1wW} U_1 + h_{1wE} h_{EW} A_{gE}) \frac{A_b}{2}$$

$$C'_1 = \alpha_g I_{SW} A_{gE} + U_{cgaW} A_{gE} T_a A_{gW}$$

$$C'_2 = \alpha_g I_{SE} A_{gE} A_{gW} h_{EW} + U_{cgaE} A_{gE} T_a A_{gW}$$

$$D_1 = m_f C_f \left(K_m^N - \frac{(1 - e^{-z})}{(1 - e^{-z} K_p^N)} \right)$$

$$D_2 = \left[\frac{(AF_R(\alpha\tau))1(1 - K_p^N)}{(1 - K_p)} \left(1 - \frac{e^{-z}}{(1 - e^{-z} K_p^N)} \right) \right]$$

$$D_3 = \left[\frac{(AF_R(UL))1(1 - K_p^N)}{(1 - K_p)} \left(1 - \frac{e^{-z}}{(1 - e^{-z} K_p^N)} \right) \right]$$

$$E_1 = U_{gaE} [h_{EW} + h_{1bwW} \left(\frac{A_b}{2A_{gW}} \right)]$$

$$E'_1 = U_{gaW} [h_{EW} + h_{1bwE} \left(\frac{A_b}{2A_{gE}} \right)]$$

$$E_2 = U_{gaW} (h_{EW} + U_{gaE}) A_{gE} A_{gW}$$

$$E'_2 = U_{gaE} (h_{EW} + U_{gaW}) A_{gE} A_{gW}$$

$$H_1 = (U_{gaE} + U_{gaW}) h_{1bwW} h_{1bwE} \left(\frac{A_b}{2} \right)$$

$$H_2 = A_{gE} A_{gW} h_{EW} (U_{gaE} h_{1bwE} + U_{gaW} h_{1bwW})$$

$$H_3 = A_{gE} A_{gW} U_{gaE} U_{gaW} (h_{1bwE} + h_{1bwW})$$

$$H_4 = A_{gE} A_{gW} h_{EW} (U_{gaE} h_{1bwW} + U_{gaW} h_{1bwE})$$

$$H'_{11} = H_1 + H_2 + H_3 + H_4$$

$$H'_{33} = U_b A_b - D_1$$

$$H'_{44} = U_b A_b + D_3$$

$$K_{1E} = [h_{1bwW} \left(\frac{A_b}{2A_{gW}} \right) + h_{EW} \left(1 + \frac{1}{h_{1bwE}} \right) + U_{gaW}] A_{gW} \alpha_g A_{gE} h_{1bwE}$$

$$K_{1W} = [h_{1bwE} \left(\frac{A_b}{2A_{gE}} \right) + h_{EW} \left(1 + \frac{1}{h_{1bwW}} \right) + U_{gaE}] A_{gW} \alpha_g A_{gE} h_{1bwW}$$

$$K'_{1E} = [h_{1bwW} \left(\frac{A_b}{2} \right) + h_{EW} \left(1 + \frac{h_{1wE}}{h_{1wW}} \right)] A_{gW} + U_{gaW} A_{gW} \alpha_g A_{gE} h_{1wE}$$



$$K'_{1W} = [h_{1bwE} \left(\frac{A_b}{2}\right) + h_{EW} \left(1 + \frac{h_{1wE}}{h_{1wW}}\right) A_{gE} + U_{gaE} A_{gE}] \alpha_g A_{gW} h_{1wW}$$

$$P = U_1 U_2 - h_{EW}^2 A_{gE} A_{gW}$$

Expressions for K_k , $(A F_R(\alpha\tau))_1$ and $(A F_R U_L)_1$

used in Eq. (1) are as follows.

$$K_p = 1 - \frac{(A F_R(U_L)1)}{m_f c_f}$$

$$K_m = 1 - \frac{(A_{rm} F_{Rm}(U_L)m)}{m_f c_f}$$

$$U_{tca} = \left[\frac{1}{h_o} + \frac{L_g}{K_g}\right]^{-1}; U_{tcp} = \left[\frac{1}{h_i} + \frac{L_g}{K_g}\right]^{-1};$$

$$h_o = 5.7 + 3.8V, \quad Wm^{-2}K^{-1}; V = 1 \text{ ms}^{-1}; h_i = 5.7, Wm^{-2}K^{-1};$$

$$U_{tpa} = \left[\frac{1}{U_{tca}} + \frac{1}{U_{tcp}}\right]^{-1} + \left[\frac{1}{h'_i} + \frac{1}{h_{pf}} + \frac{L_i}{K_i}\right]^{-1};$$

$$h'_i = 2.8 + 3V', \quad Wm^{-2}K^{-1};$$

$$U_{L1} = \frac{U_{tcp} U_{tca}}{U_{tcp} + U_{tca}}; U_{L2} = U_{L1} + U_{tpa}; U_{Lm} =$$

$$\frac{h_{pf} U_{L2}}{F' h_{pf} + U_{L2}}; U_{Lc} = \frac{h_{pf} U_{tpa}}{F' h_{pf} + U_{tpa}};$$

$$PF_1 = \frac{U_{tcp}}{U_{tcp} + U_{tca}}; PF_2 = \frac{h_{pf}}{F' h_{pf} + U_{L2}}; PF_c = \frac{h_{pf}}{F' h_{pf} + U_{tpa}};$$

$$(\alpha\tau)_{1eff} = \rho(\alpha_c - \eta_c) \tau_g \beta_c \frac{A_{am}}{A_{rm}}; (\alpha\tau)_{2eff} = \rho \alpha_p \tau_g^2 (1 - \beta_c) \frac{A_{am}}{A_{rm}};$$

$$(\alpha\tau)_{meff} = [(\alpha\tau)_{1eff} + PF_1(\alpha\tau)_{2eff}]; (\alpha\tau)_{ceff} = PF_c \rho \alpha_p \tau_g \frac{A_{ac}}{A_{rc}};$$

$$(A F_R(\alpha\tau))_1 = \left[A_c F_{Rc}(\alpha\tau)_{ceff} + PF_2(\alpha\tau)_{meff} A_m F_{Rm} \left(1 - \frac{A_c F_{Rc} U_{Lc}}{m_f c_f}\right) \right];$$

$$(A F_R U_L)_1 = \left[A_c F_{Rc} U_{Lc} + A_m F_{Rm} U_{Lm} + A_m F_{Rm} U_{Lm} \left(1 - \frac{A_c F_{Rc} U_{Lc}}{m_f c_f}\right) \right]$$

$$A_{rm} = b_r L_{rm}; A_{am} = b_o L_{am};$$

$$A_c F_{Rc} = \frac{m_f c_f}{U_{Lc}} \left[1 - \exp\left(\frac{-F' U_{Lc} A_c}{m_f c_f}\right) \right];$$

$$A_m F_{Rm} = \frac{m_f c_f}{U_{Lm}} \left[1 - \exp\left(\frac{-F' U_{Lm} A_m}{m_f c_f}\right) \right];$$

$$z = \frac{2\pi r_{11} U_L}{m_f c_f}$$

CONFLICT OF INTEREST

The authors hereby declare that there is no conflict of interest.

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